

substrates bearing base-sensitive functional groups. Alternatively, a combination of reducing reagents (NaCNBH_3 , $\text{Et}_3\text{-SiH}$, and BH_3) with Lewis acids (AlCl_3 , $\text{BF}_3\cdot\text{OEt}_2$, TMSCl , PhBCl_2 , and Bu_2BOTf) are also described in the literature.^{3a,b,e,f,6} Since these traditional Lewis acids are extremely moisture-sensitive and often have to be used in stoichiometric amounts in the reactions, there is a need to search for new catalysts that are applicable to a wide range of substrates.

Metal(III) trifluoromethanesulfonates ($\text{M} = \text{lanthanide elements, Sc, In, and Bi}$) and oxavanadium(IV) trifluoromethanesulfonate [$\text{V}(\text{O})(\text{OTf})_2$] are water-stable, reusable, and valuable Lewis acids that have been used as efficient catalysts in a variety of reactions.^{7–13} To the best of our knowledge to date, these metal trifluoromethanesulfonates have not been employed in conjunction with a reducing

agent. We have explored herein their catalytic activities in the BH_3 -reductive cleavage of benzylidene acetals.

Methyl 4,6-*O*-benzylidene-2,3-di-*O*-benzyl- α -D-glucopyranoside **1** was selected as the model compound for our studies. Treatment of **1** with $\text{BH}_3\cdot\text{THF}$ in the presence of 15 mol % $\text{M}(\text{OTf})_n$ gave the corresponding 6-alcohol **2** in excellent yields. The reaction conditions and results are illustrated in Table 1. Initially (entry 1), when the benzylidene

Table 1. $\text{M}(\text{OTf})_n$ -Catalyzed BH_3 -Reductive Ring Opening of the 4,6-*O*-Benzylidene Acetal **1** at 06

entry	$\text{M}(\text{OTf})_n$	T ($^{\circ}\text{C}$)	t (h)	yield (%)
1	$\text{V}(\text{O})(\text{OTf})_2$	0	27	87
2	$\text{V}(\text{O})(\text{OTf})_2$	rt	3	94
3	$\text{Sc}(\text{OTf})_3$	rt	5	93
4	$\text{Pr}(\text{OTf})_3$	rt	5	91
5	$\text{Nd}(\text{OTf})_3$	rt	4	94
6	$\text{Sm}(\text{OTf})_3$	rt	3.5	92
7	$\text{Eu}(\text{OTf})_3$	rt	3	87
8	$\text{Gd}(\text{OTf})_3$	rt	3	88

acetal ring was opened at 0 $^{\circ}\text{C}$, it took a prolonged time for completion and a small amount of acetal-hydrolyzed product was also detected along with the expected product **2** (87%), whereas conducting the same transformation at ambient temperature resulted in formation of **2** (94%) in much shorter reaction time. Interestingly, all $\text{M}(\text{OTf})_n$ were appropriate as catalysts to carry out the bond cleavages at room temperature in ca. 3–5 h (entries 2–8).

On the basis of our new findings, the applications of $\text{V}(\text{O})(\text{OTf})_2$ ^{9d} in a variety of 4,6-*O*-benzylidene-D-hexopyranosides are studied further (Table 2). To test the compatibility of different protecting groups, the D-glucoside derivatives **3** and **5** were examined (entries 1 and 2), and the alcohols **4** and **6** were isolated in 83% and 92% yields, respectively. The reactions worked equally well with the β -D-glucoside (entries 3–5), α -D-galactoside (entries 6 and 7), and α -D-mannopyranosyl sugars (entry 8). It is evident from the above results that the electron-donating protecting groups in the substrate speed up the reaction while electron-withdrawing groups slow it down. Obviously, the reaction rate is concerned with the nucleophilicity of the 6-oxygen atom for the Lewis acid catalyst. Finally, to extend this reaction in the D-glucosamine derivative **19**,¹⁴ the corresponding 6-alcohol **20** was successfully obtained in 84% yield (entry 9). Its absolute configuration was determined through the X-ray single-crystal analysis (see Supporting Information).

Iminosugars, which are polyhydroxylated nitrogen-containing heterocycles, have been attracting the interest of

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Table 2. Borane-Reductive Ring Opening of Various 4,6-*O*-Benzylidene-D-hexopyranosides at Room Temperature in the Presence of 15 Mol % V(O)(OTf)₂ as Catalyst

entry	acetal	t (h)	product	yield (%)
1		12		83
2		16		92
3		3		86
4		7.5		90
5		1		86
6		3		88
7		18		74
8		3		91
9		3		90

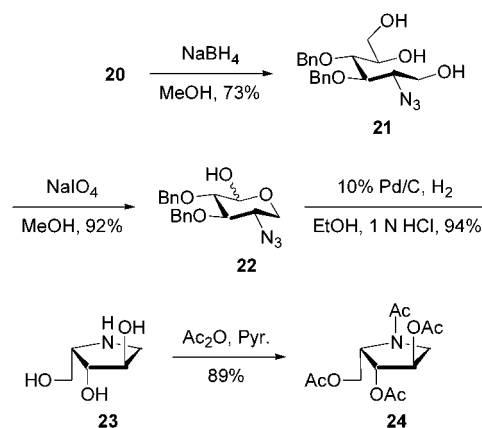
synthetic chemists as a result of their potent glycosidase inhibition, antiviral activities, and anticancer properties.¹⁵ As a part of our ongoing research program to synthesize the biologically important and rare L-form sugars,¹⁶ we proceeded to prepare 1,4-dideoxy-1,4-imino-L-xylitol **23**,¹⁷ a potential inhibitor of glycosidases.

With the key synthon **20** in hand, the synthesis of **23** was efficiently carried out in three steps (Scheme 1). Reduction

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Scheme 1



of **20** with sodium borohydride led to the triol **21** (73%), which was subjected to NaIO₄ oxidative cleavage of carbon–carbon single bond between the vicinal hydroxyl groups to furnish the cyclic hemiacetal **22** (92%). Further reduction of **22** under hydrogenation conditions afforded the target molecule **23** in 94% yield. Comparison of our data of its peracetylated derivative **24** with the literature report^{17f} revealed identity with respect to ¹H and ¹³C spectra.

In conclusion, we have successfully developed a M(OTf)_n-catalyzed regioselective BH₃-reductive ring opening of the 4,6-*O*-benzylidene-D-hexopyranosides to the corresponding 6-alcohols in excellent yields at room temperature and applied the key synthon **20** to synthesize 1,4-dideoxy-1,4-imino-L-xylitol **23** in three straightforward steps. This representative example demonstrates the inherent synthetic potential of this methodology. The efforts to prepare other biologically important iminocyclitols are currently underway.

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Supporting Information Available: Experimental procedures and characterization data for all new compounds and X-ray structural information for compound **20**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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